

# Particle Measurement for Iron Transport and Iron Analysis

2015 Air Cooled Condenser Users Group

Gettysburg, PA

September 21 - 24, 2015

Ken Kuruc



# OBJECTIVES

- Introduction to FAC Monitoring
- Alternatives for monitoring corrosion product transport in the steam cycle
  - On-line surrogate method (suspended iron)
  - Grab sample method (total iron)
  - Millipore filter pad (suspended)
- Data from plants with ACCs

# Overview

## Flow Accelerated Corrosion (FAC):

- a chemical and mechanical corrosion mechanism which steadily erodes the inside of piping and other carbon steel components.
- identified as the cause of several power plant accidents & deaths
- is extremely expensive in terms of mechanical failures
- is particularly problematic for plants using air cooled condensers

# FAC Overview

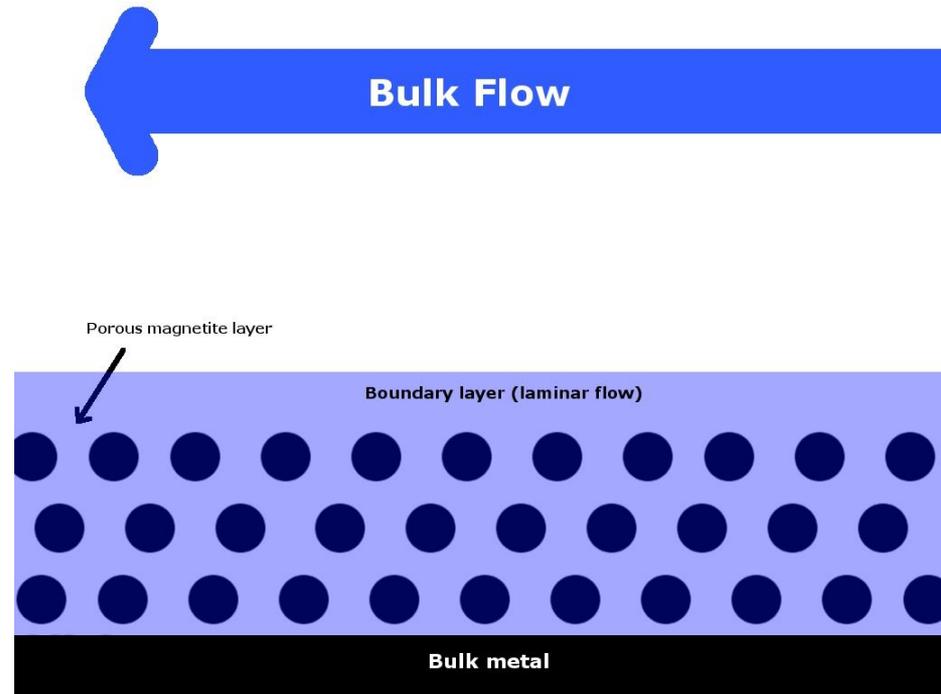
## Protective Magnetite

**Reducing** with laminar flow

Soluble ferrous iron is transported from the bulk metal to the boundary layer

Ferrous iron creates a porous protective layer of particulate magnetite,  $\text{Fe}_3\text{O}_4$ .

Magnetite layer protects the bulk metal



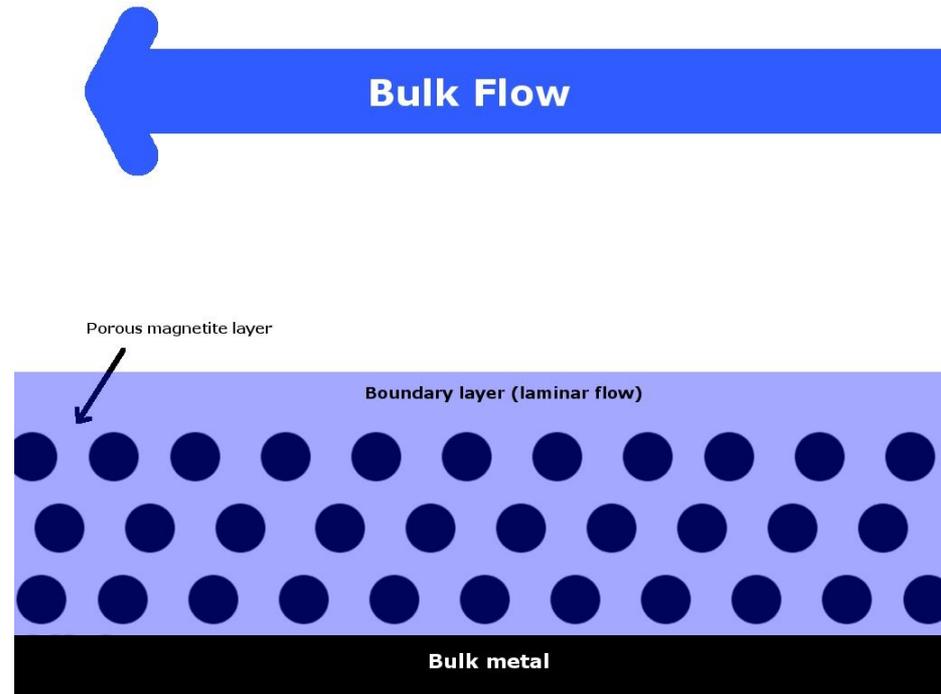
# FAC Overview

## Protective Hematite

**Oxidizing** conditions

Ferrous iron ions oxidize & combine with oxygen to form hematite

Hematite particles distribute throughout the porous magnetite lattice forming a strong protective layer



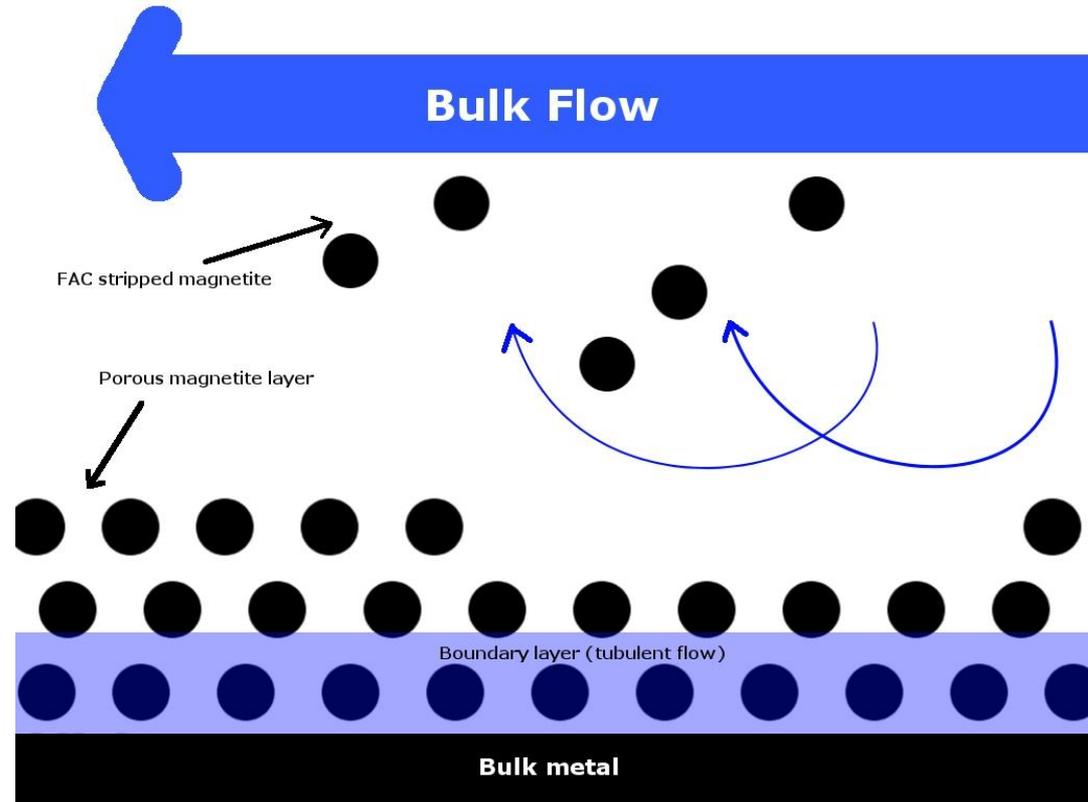
# FAC Overview

## Magnetite Stripping

Turbulent flow strips the protective magnetite layer from the metal surface

Metal is thinned and weakened

Magnetite is re-deposited in other areas



# Traditional Sampling Methods

- Inductively Coupled Plasma
  - Optical Emission Spectrometry (ICP-OES)
  - Mass Spectrometry (ICP-MS)
- X-Ray Fluorescence (XRF)
- Particle Counters & Monitors
- Colorimetric (lab) – very limited

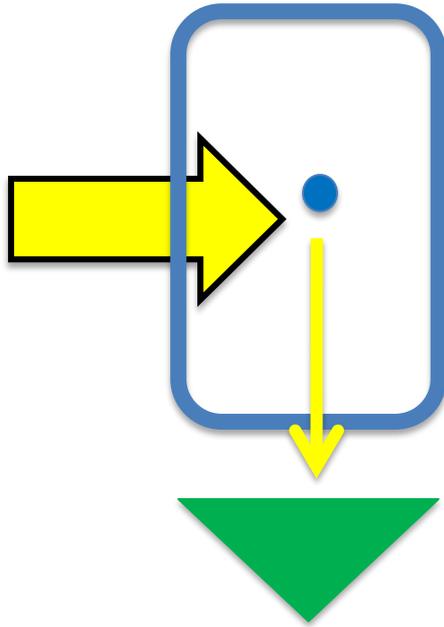


# Filter Pad Monitoring

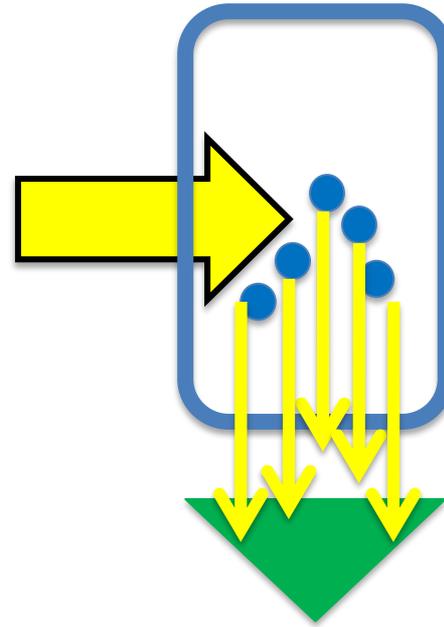
- Provides a total Fe picture *after* the fact
- Averaging produces a deceptive picture of the real corrosion environment



# Light Scattering Propagation of Submicron Particles



**Non-detectable**



**Detectable**

# Laser Nephelometry

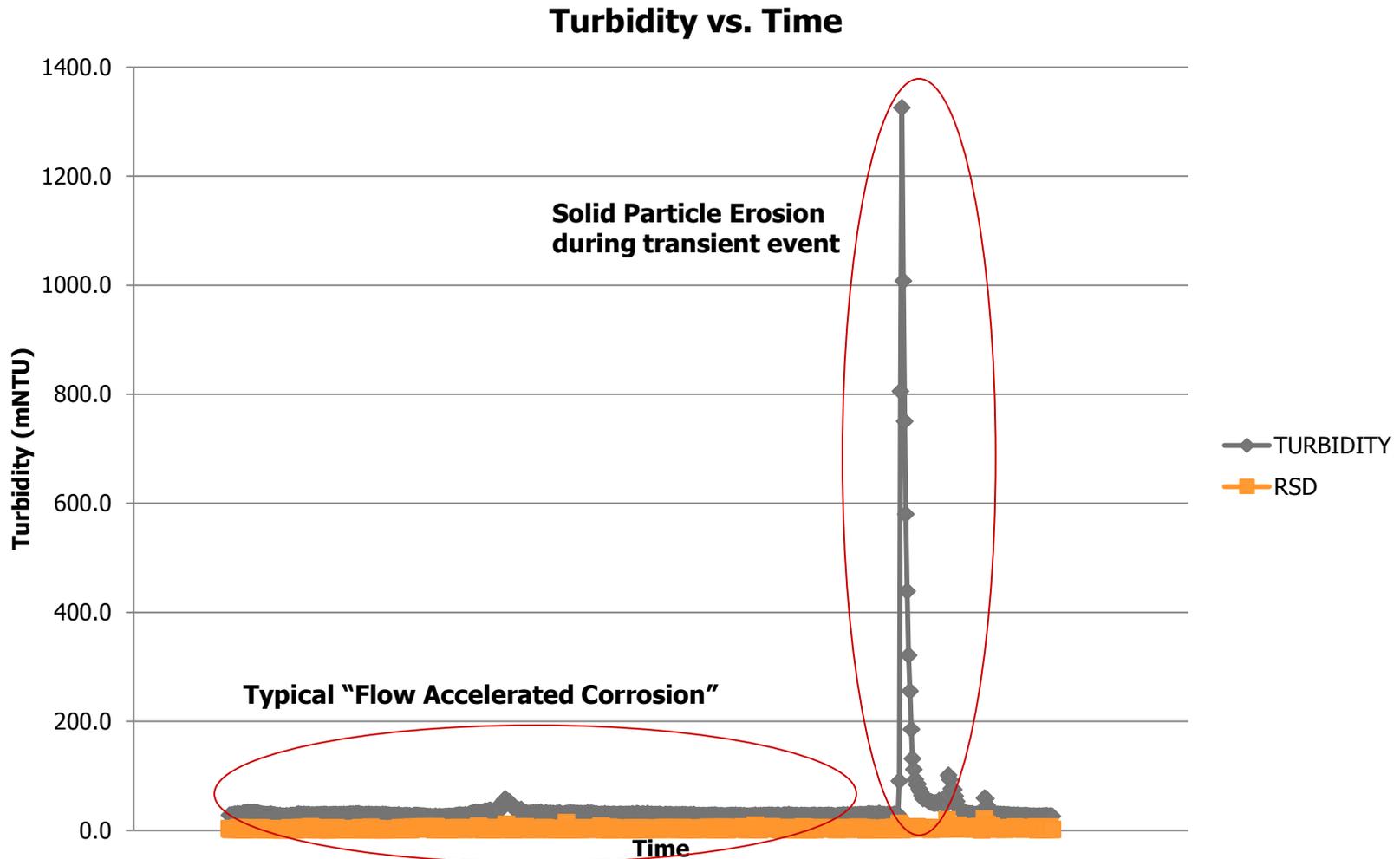
Potential solution for detection of insoluble iron:

- Theoretically 1 mNTU = 1 ppb total solids (kaolin)
- LOD is 0.3 mNTU
- Particle sizes <1 micron are detected when concentration of particles is such that the sum reflectance can increase photodetector response above 7 mNTU.

**A reference bench method (FerroZine) has been developed to support the on-line research**

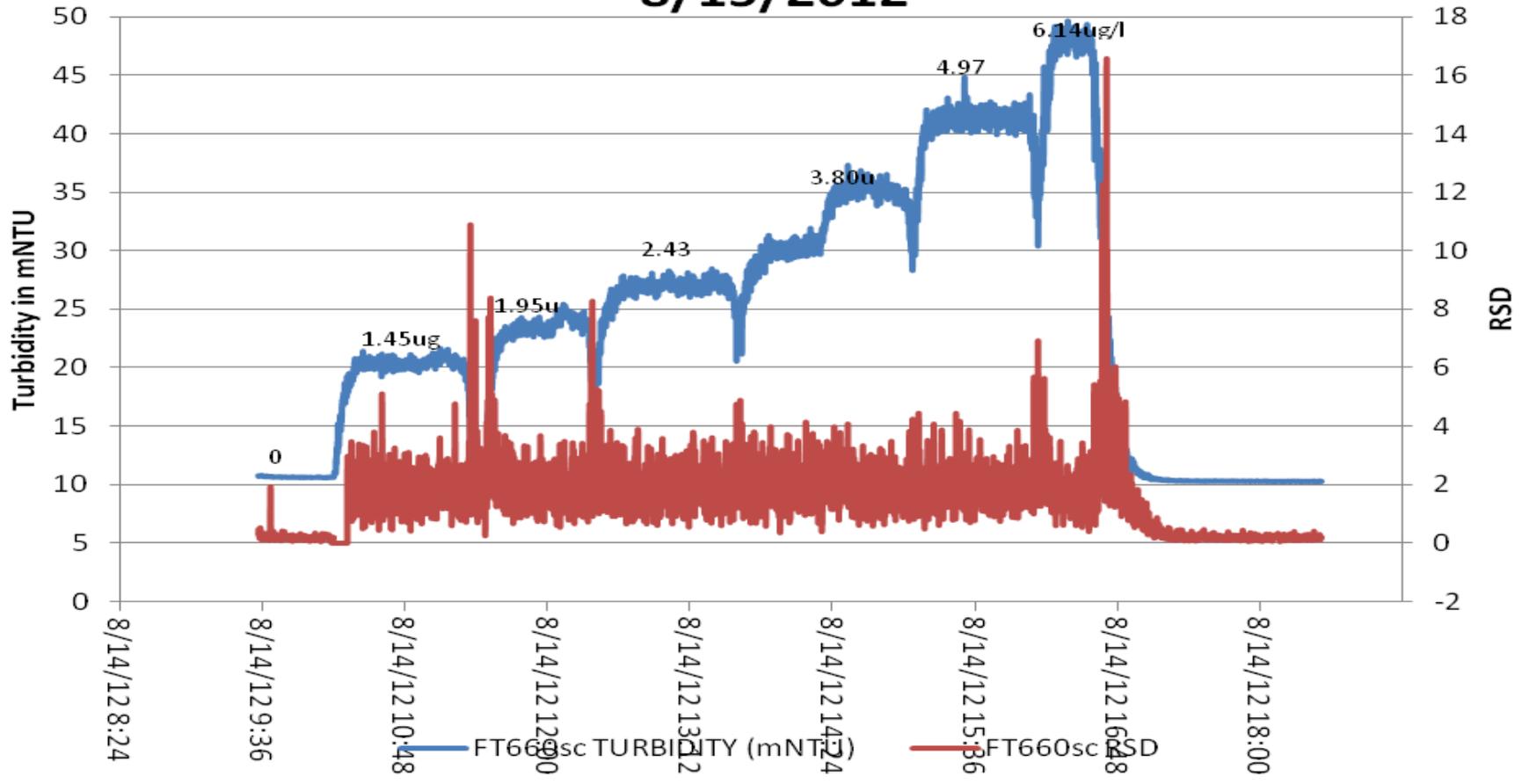


# Corrosion Product Monitoring



# Magnetite

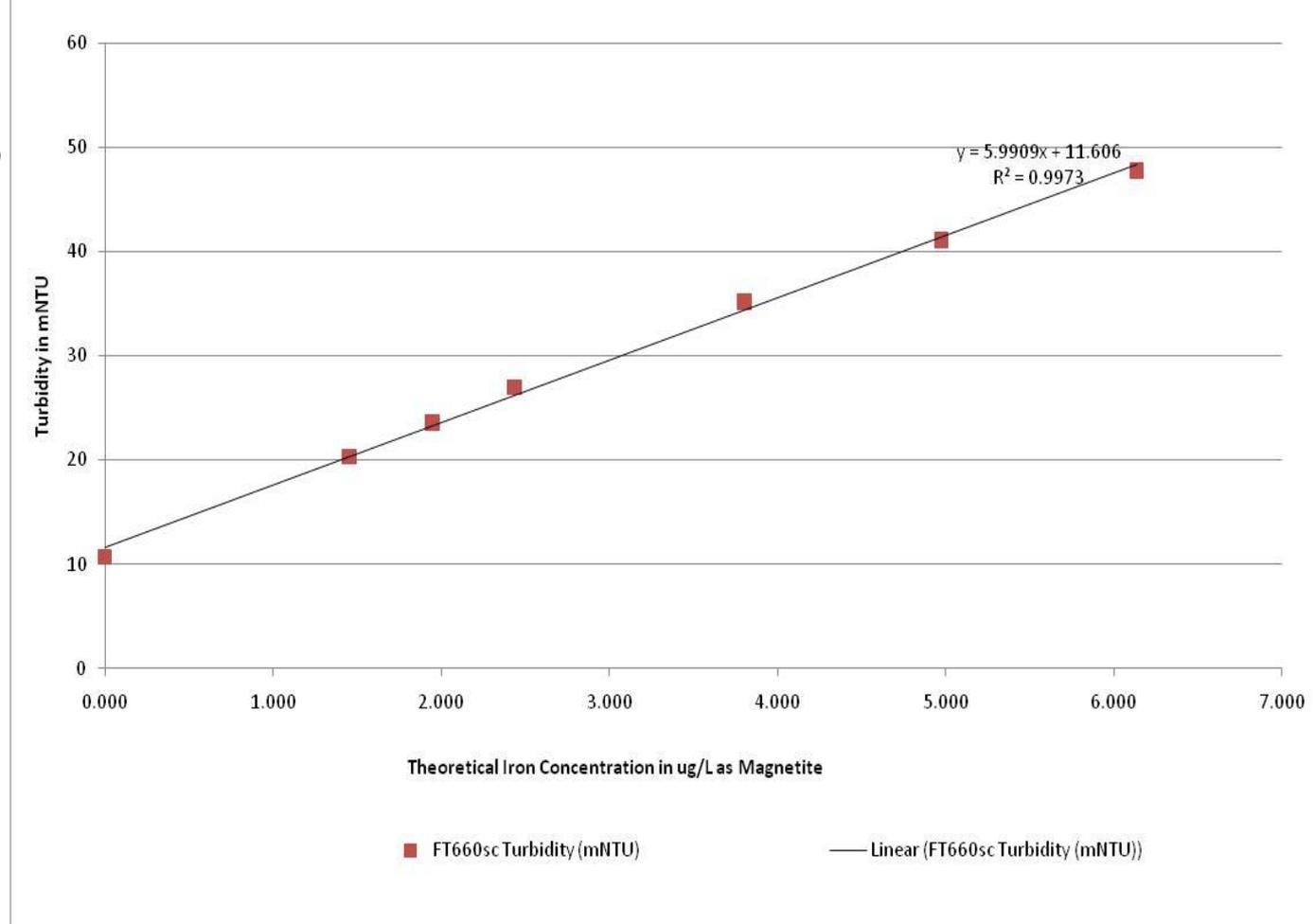
## FT660sc - Magnetite Correlation to Turbidity 8/15/2012



# Correlation Data – MAGNETITE

## Insoluble Iron Detection

Correlation of Magnetite to Laser Nephelometric Response for Axicon and the FT660sc - 8/15/12



### Laser Nephelometry Performance (Magnetite)

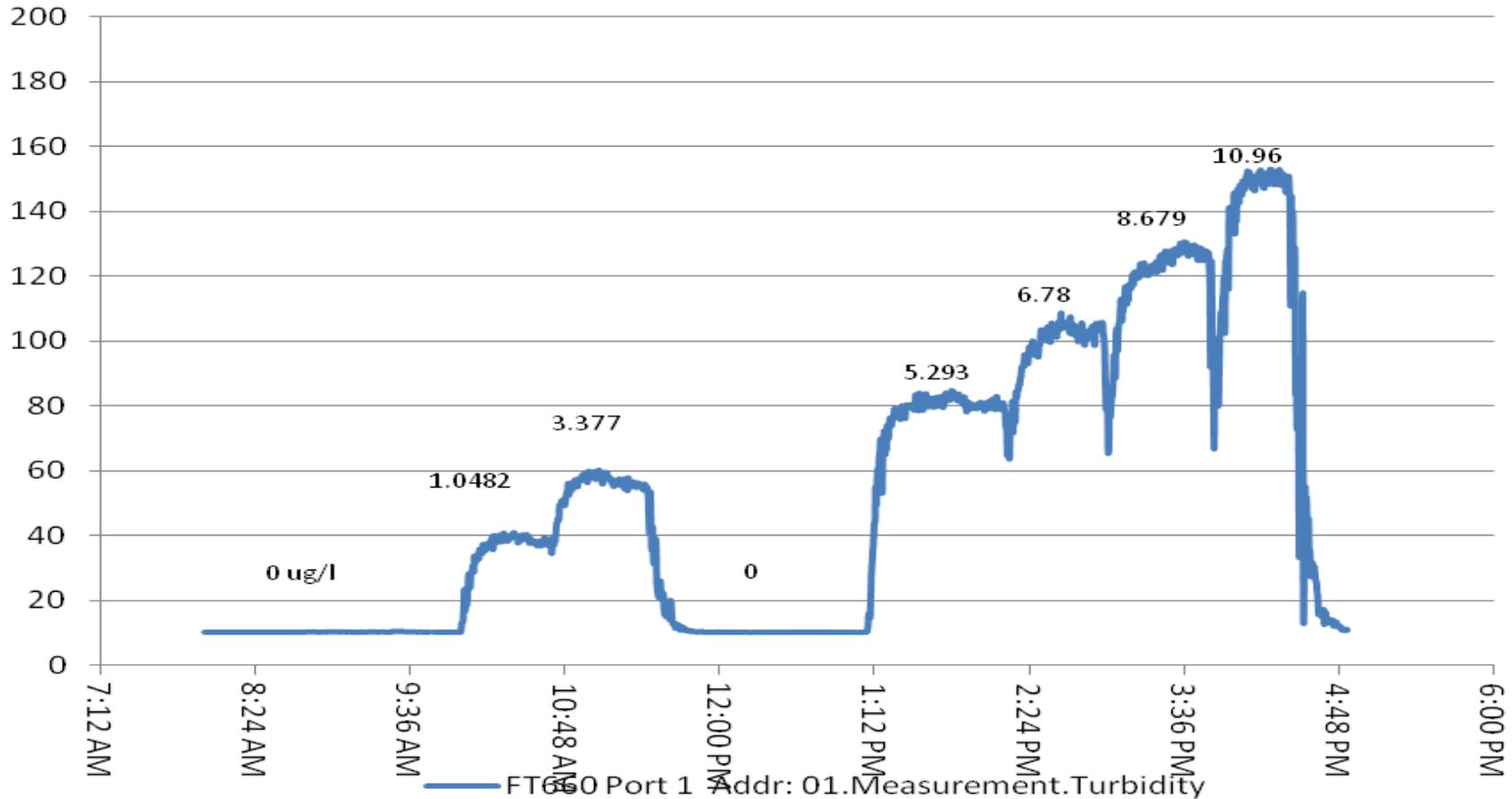
An increase of 1-ppb iron results in an increase of 6 mNTU

Linear response curve

Detection limit 50 ppt range

# Hematite

## FT660 - Hematite Correlation to Turbidity



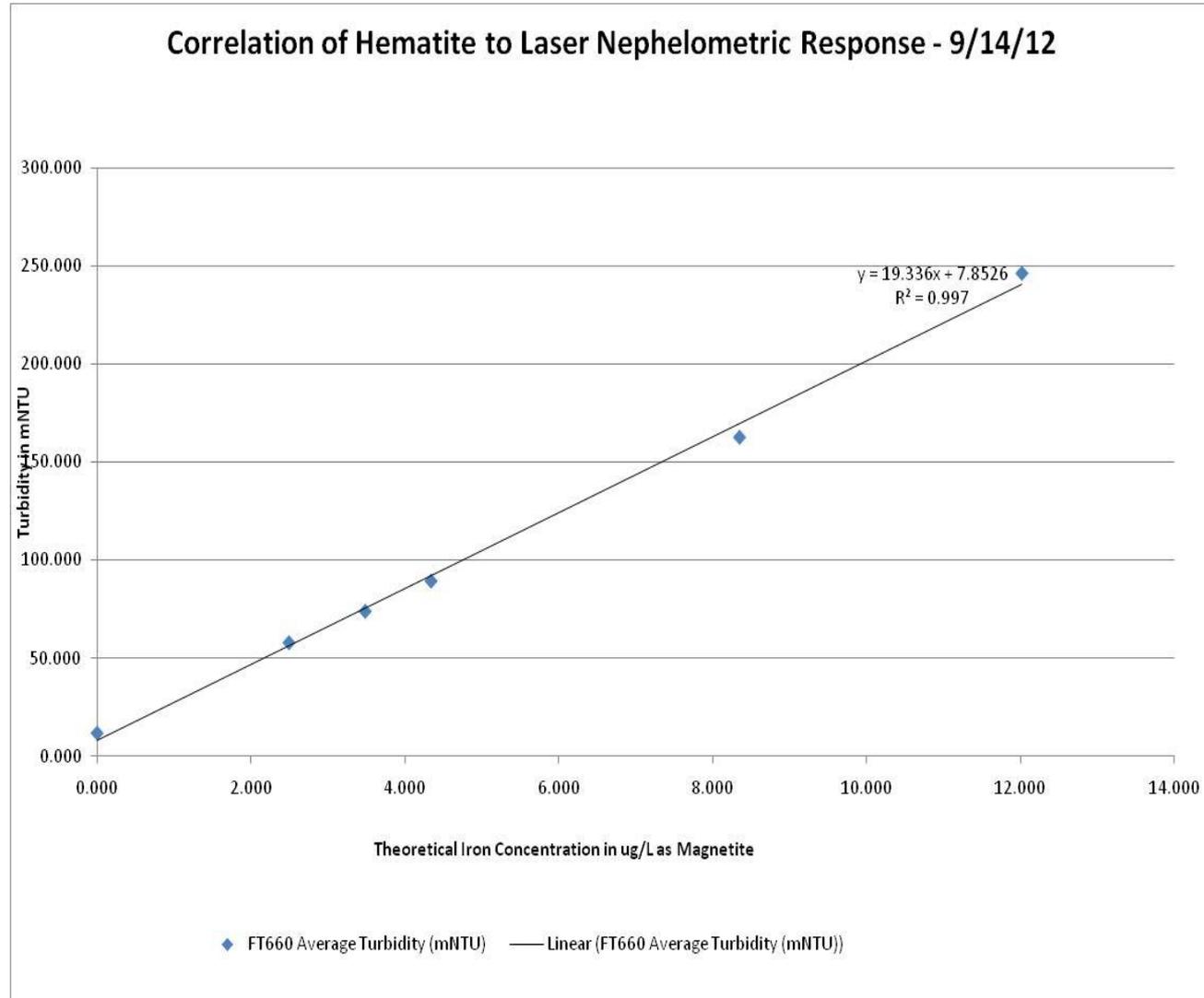
# Correlation Data – HEMATITE

## Insoluble Iron Detection

### Laser Nephelometry Performance (Hematite)

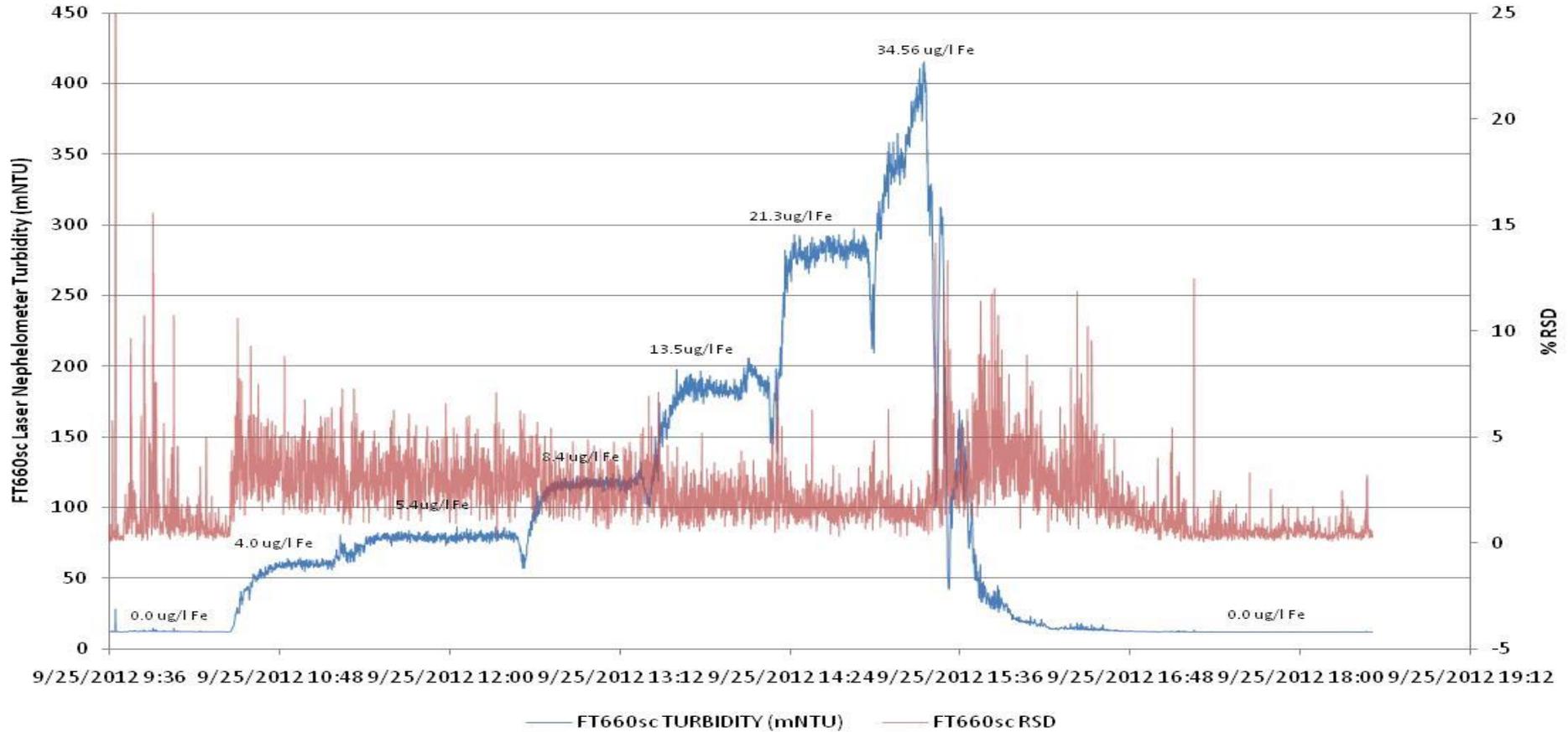
An increase of 1-ppb iron results in an increase of 20 mNTU Linear response curve.

Detection limit 15 ppt range



# 50:50 Mixture

## Turbidimetric Measurement of Hematite and Magnetite 50-50 Fraction by Mass - 9/25/2012



# Correlation Data – 50:50

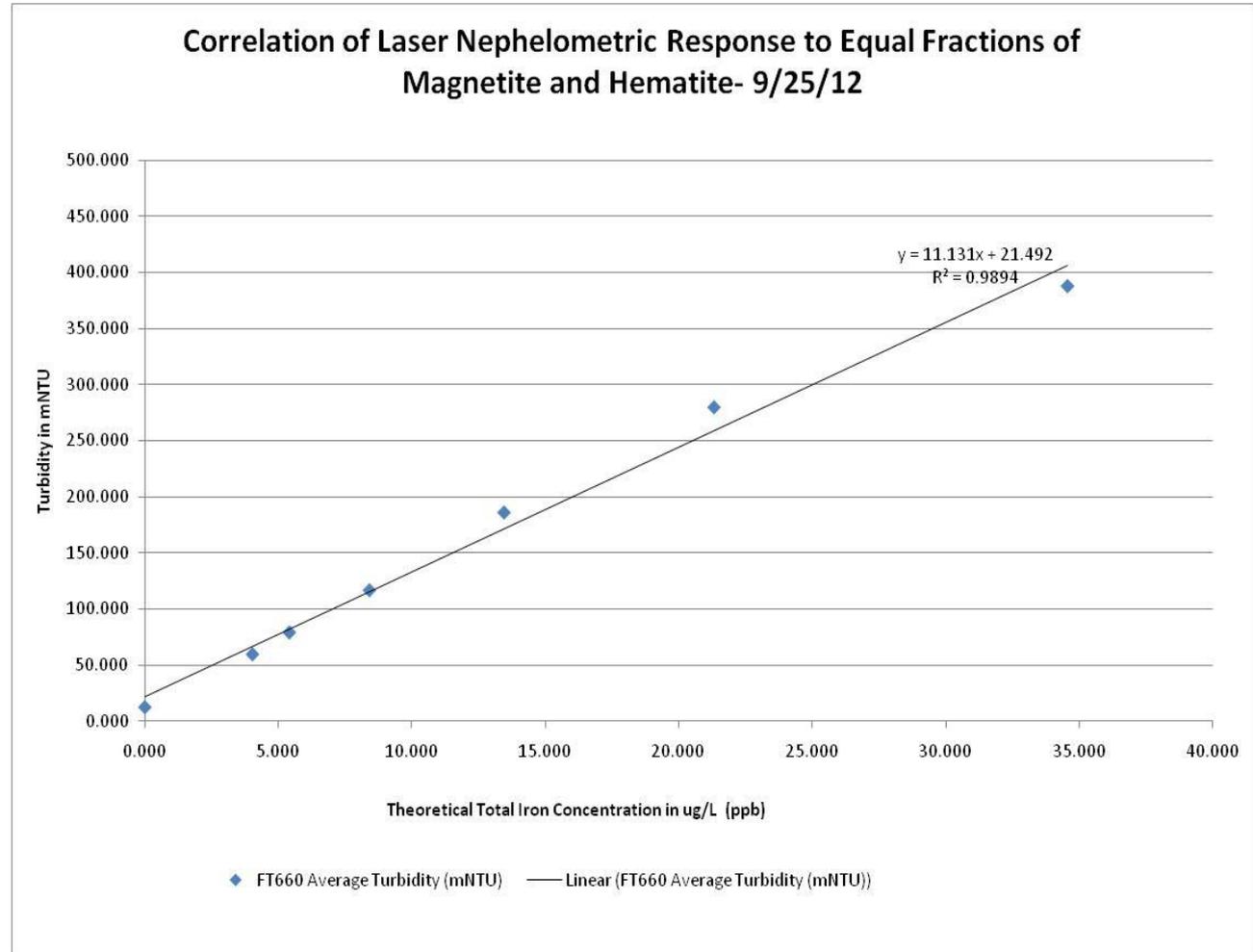
## Insoluble Iron Detection – Magnetite – Hematite 50:50 Mass Fraction

### Laser Nephelometry Performance

An increase of 1-ppb iron results in an increase of 11 mNTU (FT660)

Linear over the tested range

Detection limit 27 ppt range as total iron



# Determination of Total Iron

## (Bench Method)

- Analytical Objective
  - Measure total iron in the plant at  $<2.0 \mu\text{g/L}$
  - Analysis to be convenient to perform at plant level
    - Multiple analysts able to perform test
  - Measure soluble and total iron including particulate.
    - Ferrous and ferric
    - Magnetite
    - Hematite
- Tactical Objective
  - In-plant analysis of iron
  - Correlation of iron to laser turbidity



8

26

Fe

Iron  
55.847

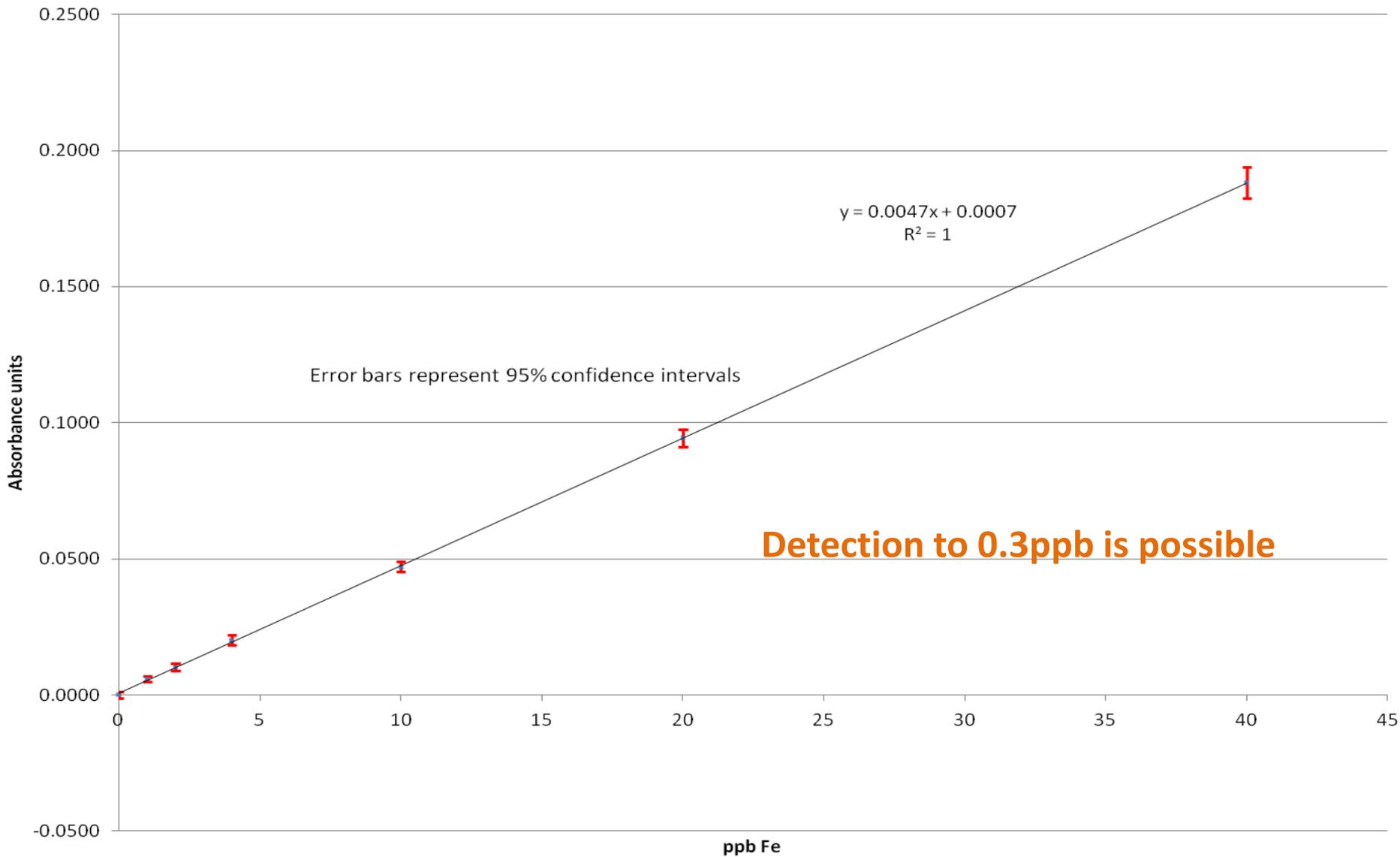
# ULR FerroZine Method

- *Combination* reagent - FerroZine + TGA
  - TGA is responsible for digestion
  - FerroZine creates the color
- Reduces *all Fe* including magnetite & hematite
- 150°C digestion for 30 mins

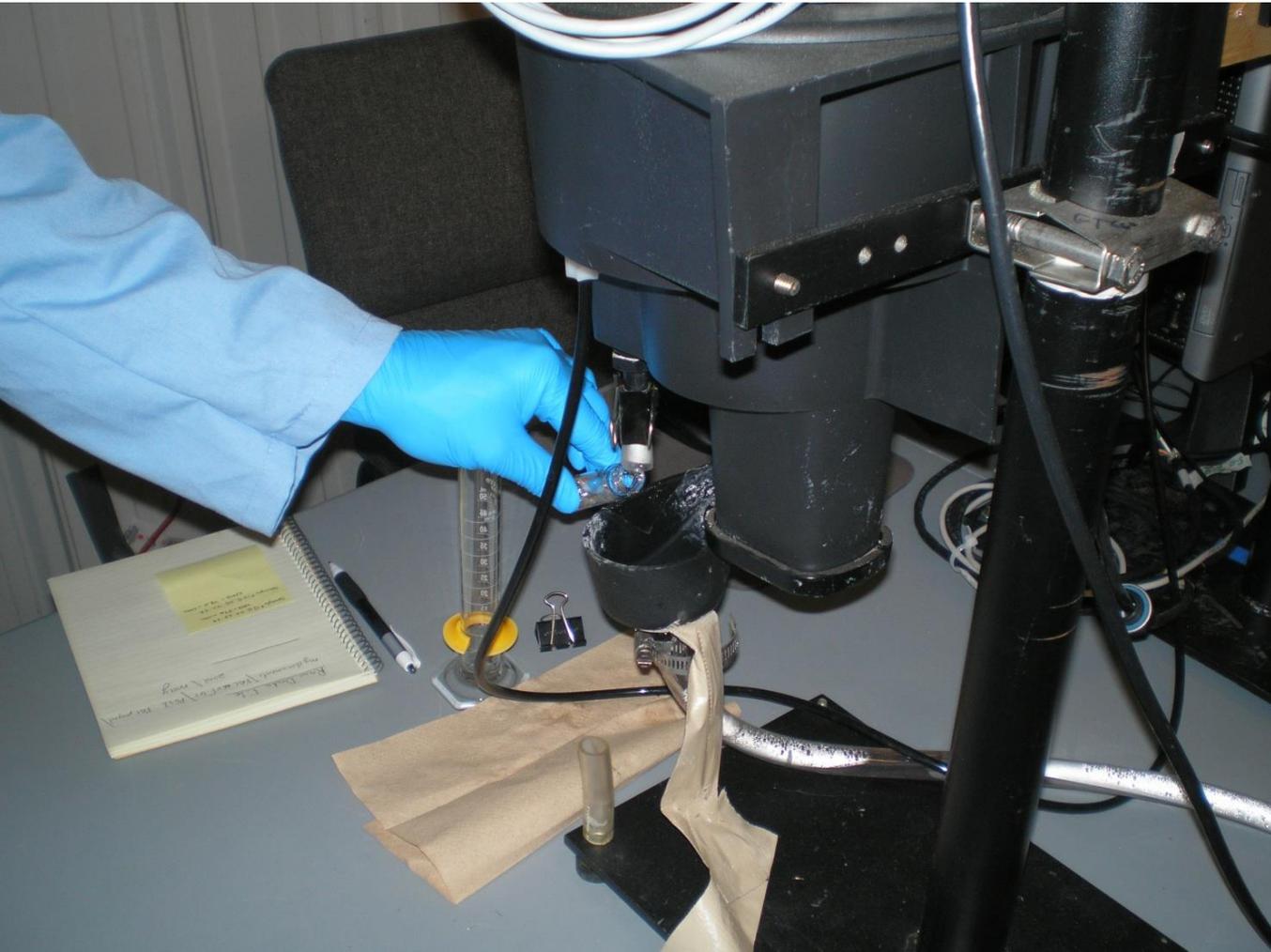
**Designed for low ppb range**

# Ultra Low Range Iron Analysis

## ULR Fe Calibration

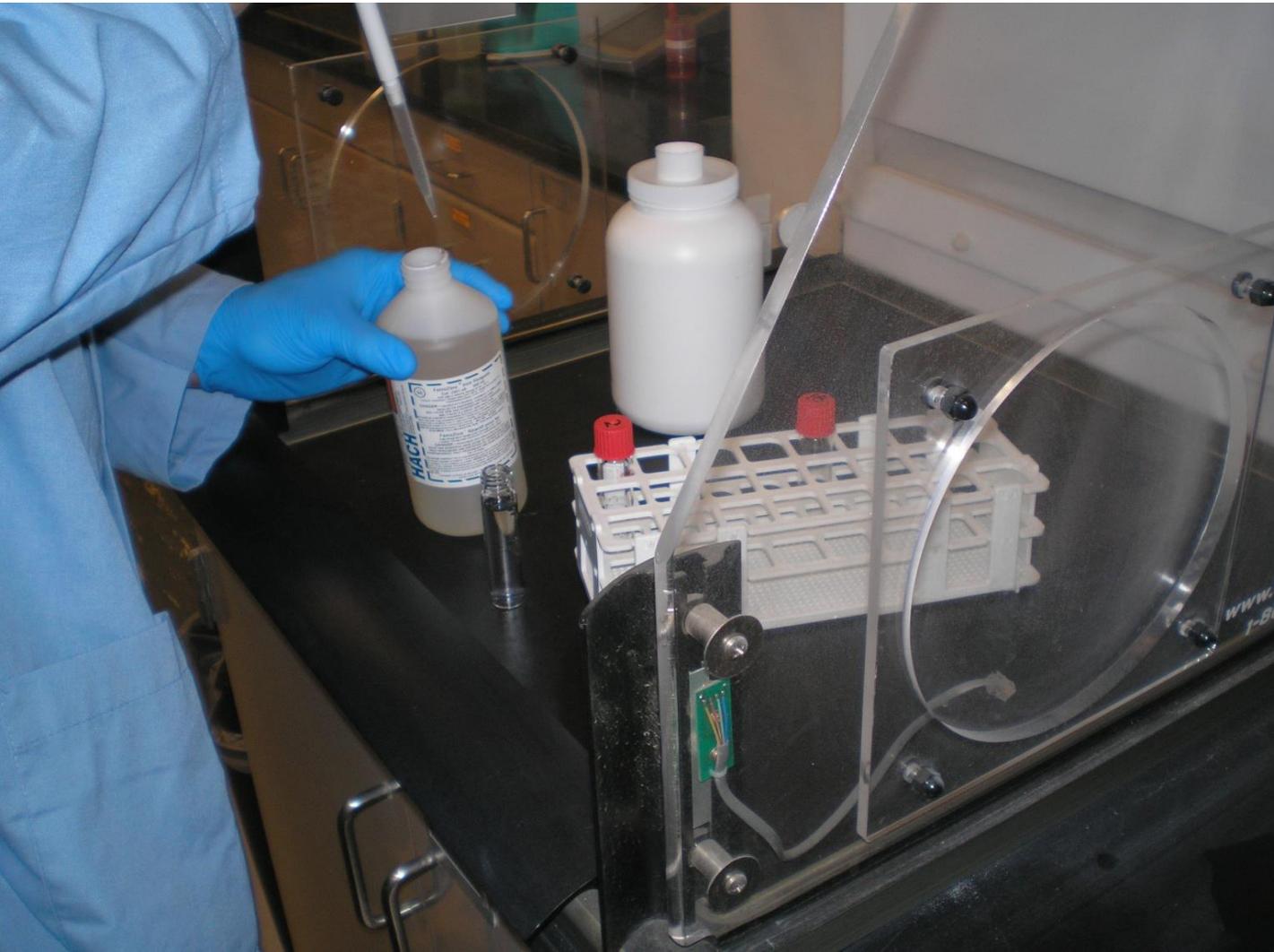


# Sample Acquisition



Attention to  
technique and  
method is  
critical

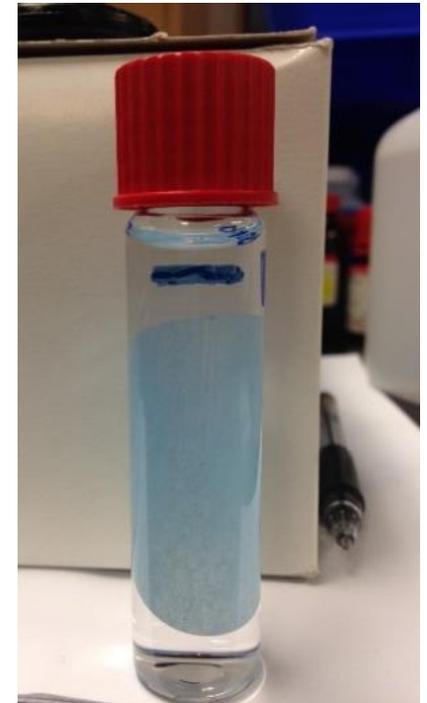
# Addition of Reagents



FerroZine +  
TGA  
(Thioglycolic  
acid)

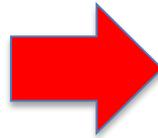
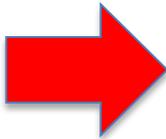
# Dissolution

Using a digestion block at 150°C for 30 minutes



# Measurement

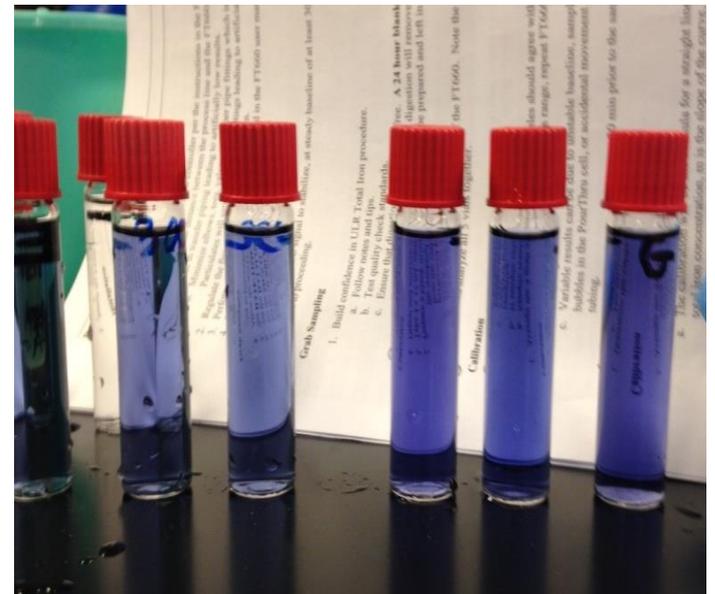
Spectrometer must have high quality optics



# Filter Pad Procedure

- Place the filter pad into vial containing 15ml of DI water
- Add 10 drops of FerroZine and digest for 30 mins at 150°C
- Dilute sample into analytical range
- Add 8 drops of FerroZine to the diluted sample
- Run in spectrophotometer
- Calculate iron on the filter pad

NOTE: ALL vials should initially be cleaned by heating 10 drops of FerroZine in 15 ml H<sub>2</sub>O for 24 hrs to extract Fe from the glassware



# Example

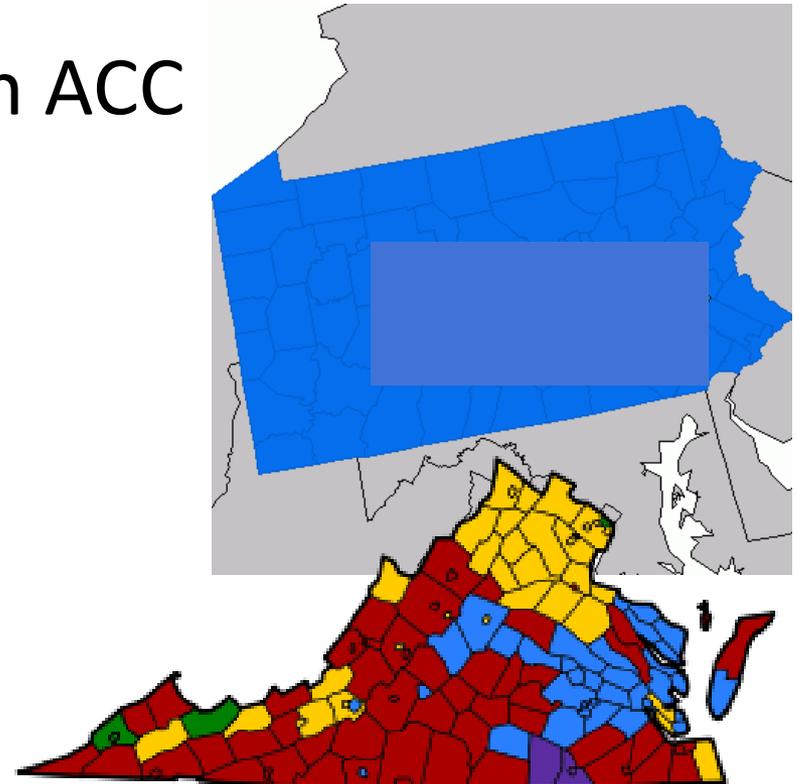
If the diluted concentration is 0.900 mg/L, and the sample was diluted 1 mL/500 mL, then the total mass of Fe on the filter pad is:

$$\frac{0.900 \frac{\text{mg}}{\text{L}} \times 500 \text{mL} \times 0.015 \text{L}}{1 \text{ mL}} = 6.75 \text{ mg}$$



# Plant Studies

- California HRSG with ACC
- Virginia HRSG with ACC
- Pennsylvania HRSG with ACC



# California Study: 5/12 – 8/12

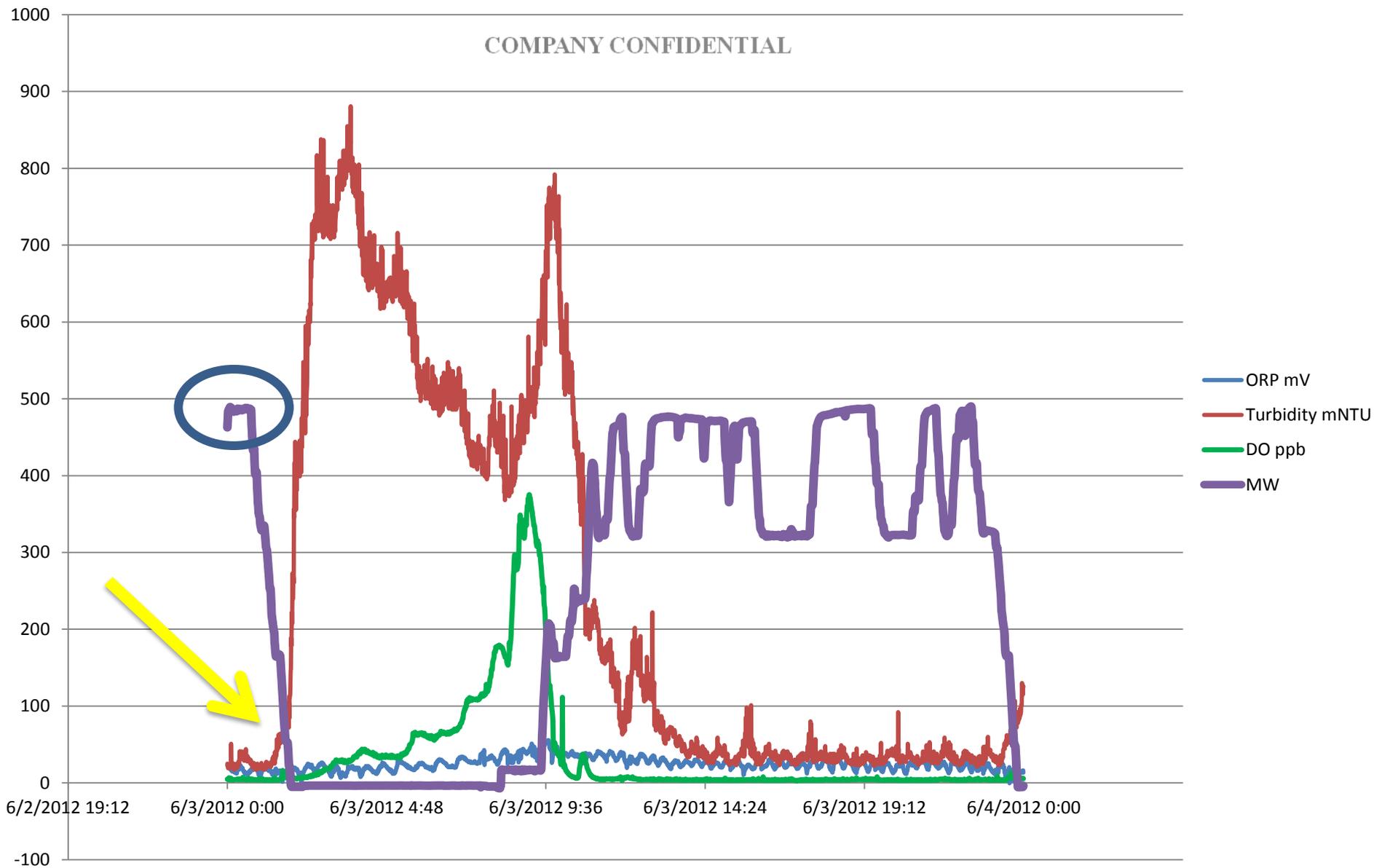
- HRSG Plant
  - 2X1
  - 600 MW
  - ACC
- Continuous Monitoring of Condensate:
  - Laser Turbidity (mNTU)
  - pH
  - ORP
  - DO
  - Temperature
- Continuous data is logged to computer every 30 seconds.



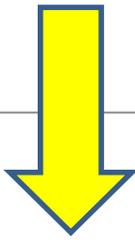


# June 3 00:00 – 24:00

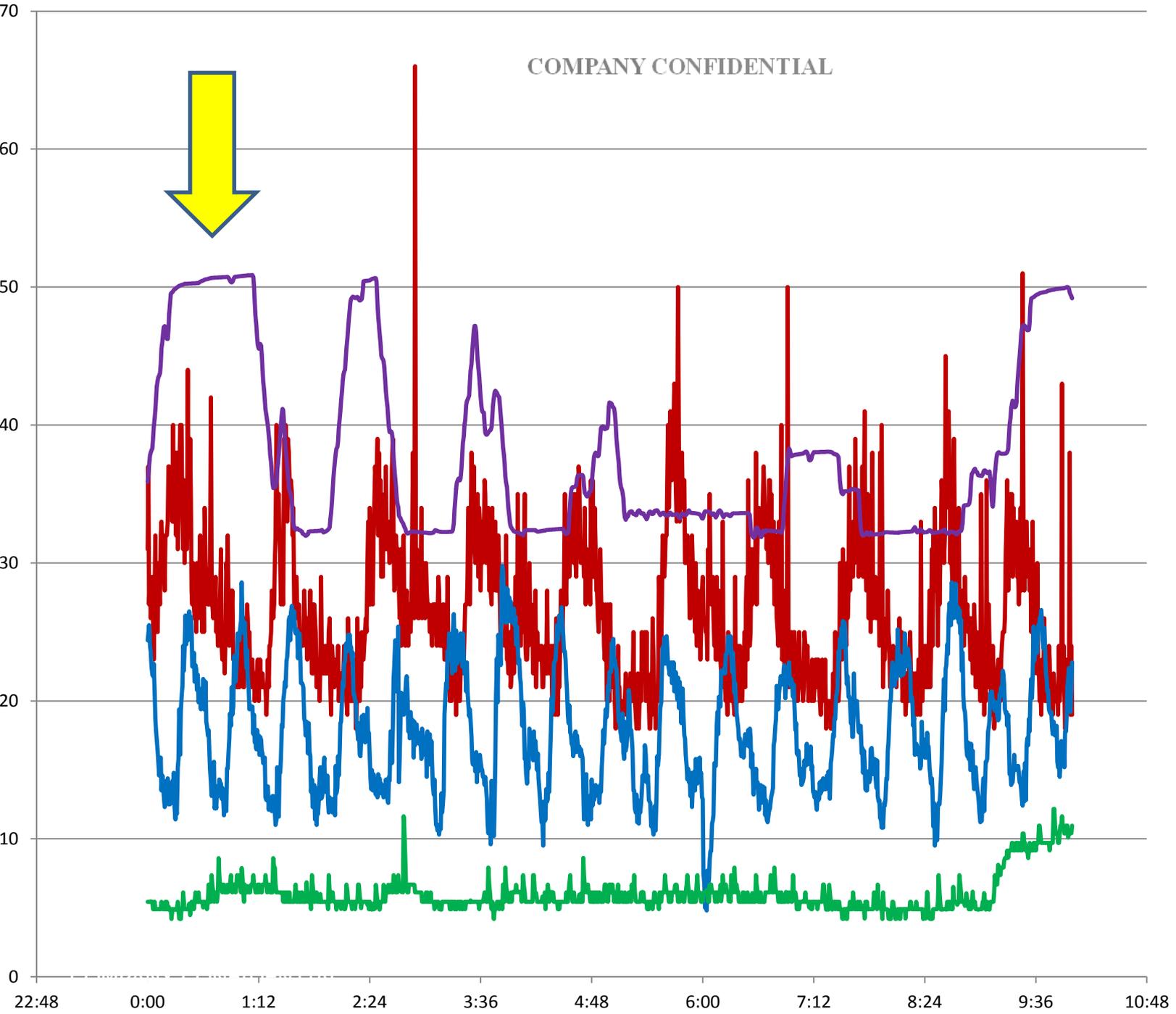
COMPANY CONFIDENTIAL



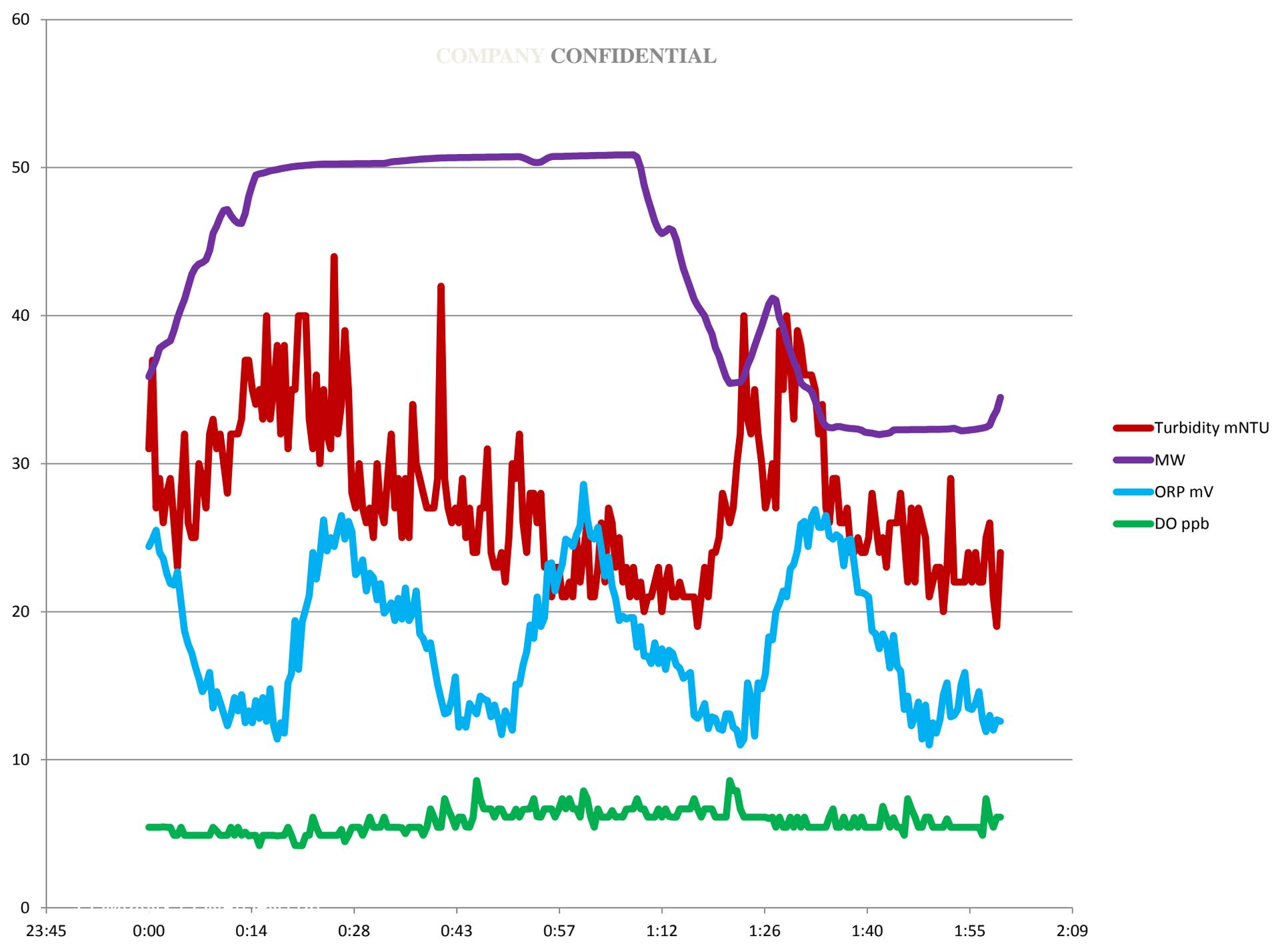
COMPANY CONFIDENTIAL



- Turbidity mNTU
- MW
- ORP mV
- DO ppb



COMPANY CONFIDENTIAL



# Virginia Study: 7/13 – 6/15

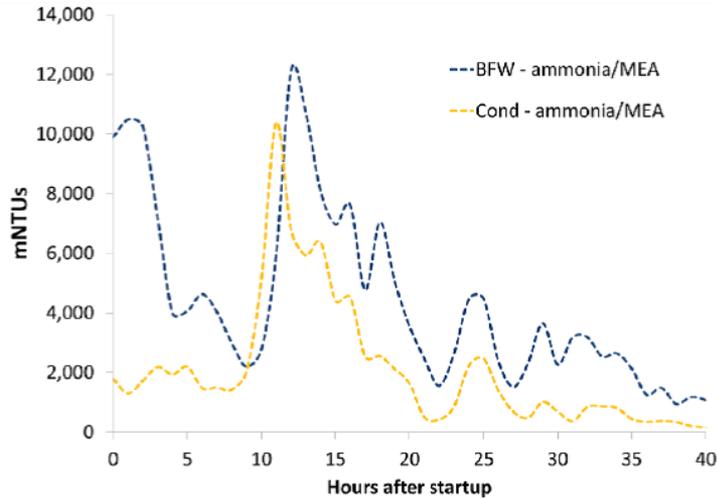
- Coal Plant
  - 600 MW
  - ACC
- Feeding Polyamine



# Cold start w/ ammonia:MEA AVT

Turbidity peaked ~12K mNTU

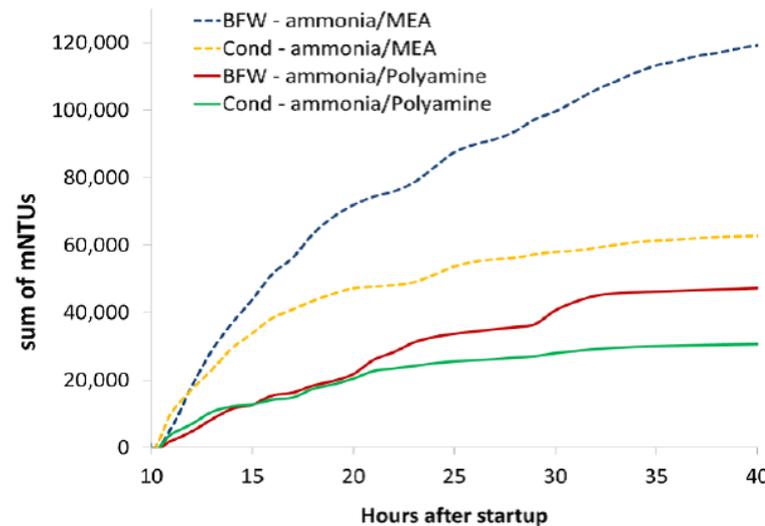
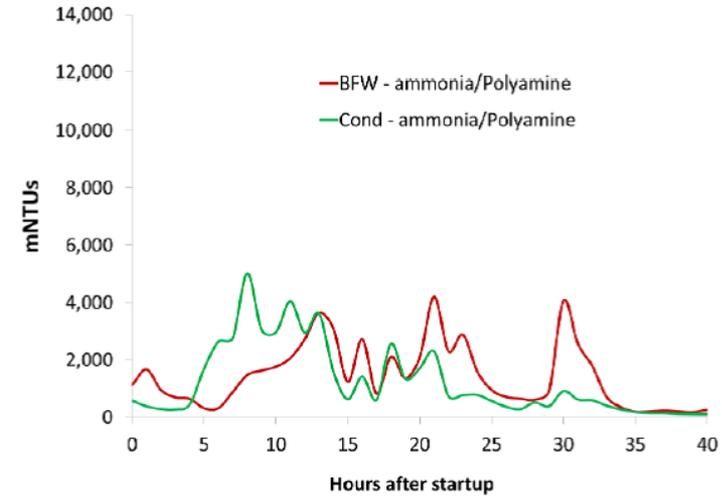
Condensate turbidity in 2-10K mNTU range



# Cold start after Polyamine addition

Turbidity peaked ~5K mNTU

Condensate turbidity in 1-5K mNTU range

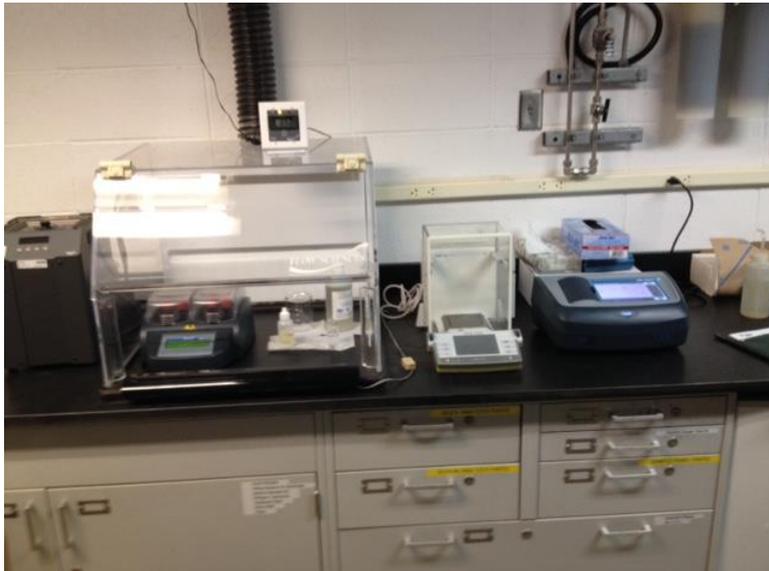


Courtesy GE Power & Water

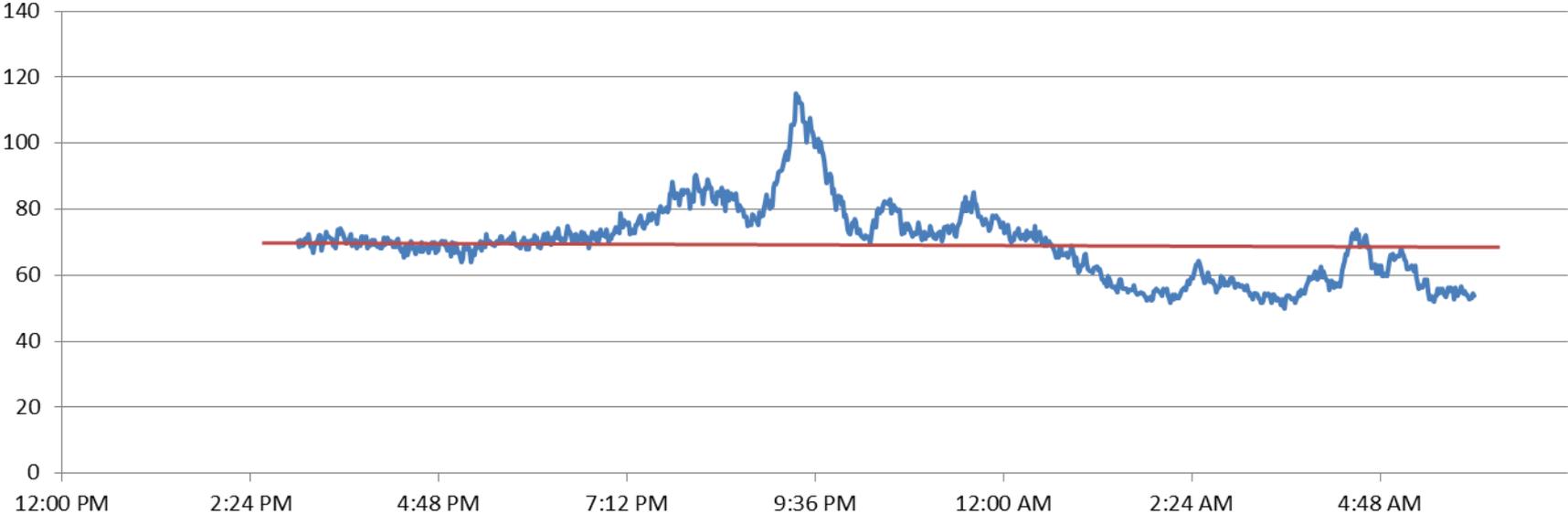


# Pennsylvania Study: 8/15 - Present

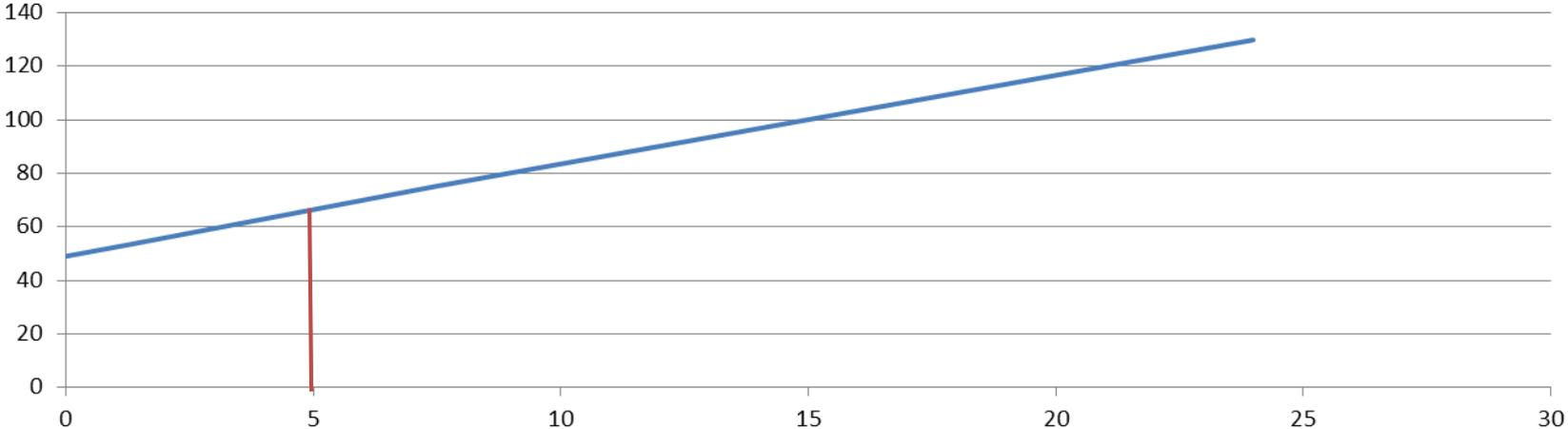
- Combined Cycle
  - 900 MW
  - ACC



# NRG Hunterstown - Condensate Iron (8/25 - 8/26)

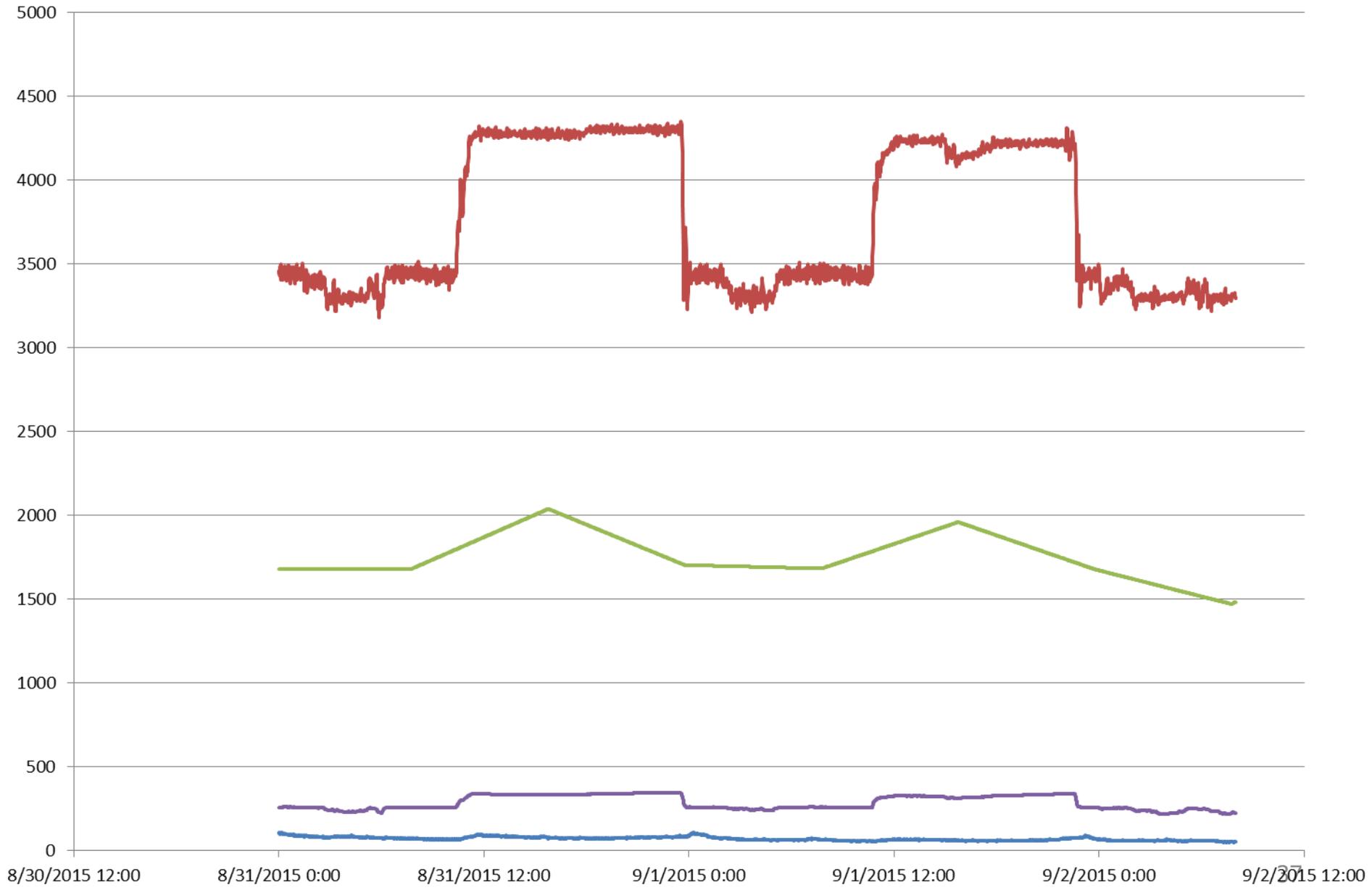


## Correlation Curve



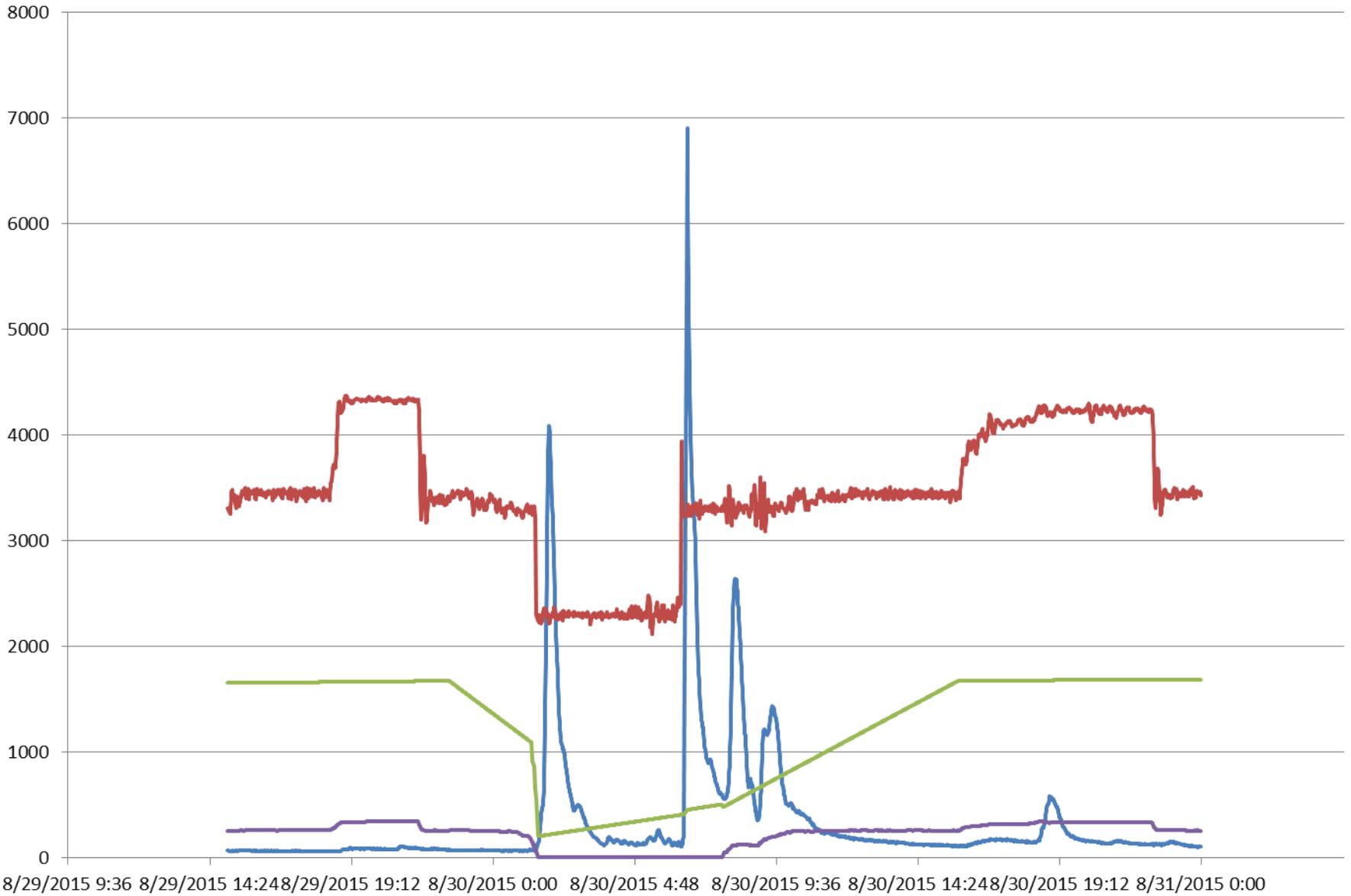
# Monday 8/31 - Wednesday 9/2

TURBIDITY    CONDENSATE FLOW IN GPM    STEAM FLOW    MW



# Saturday 3:00PM - Sunday 11:59PM

TURBIDITY    CONDENSATE FLOW IN GPM    STEAM FLOW    MW



# TAKE AWAY!

- Surrogate methods have now been validated for monitoring iron continuously
- Total iron methods have been improved for grab samples to levels down to 1 ppb



# Questions

