# **KOGAN CREEK UPDATE** 4<sup>TH</sup> ANNUAL ACC USERS GROUP MEETING – 25 – 26 SEPT 2012



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### **Presentation Overview**

- Kogan Creek Power Station History Review
- Turbine MW Degradation
- HP Turbine Blade Deposits Analysis
- Corrosion Product Sampling Deposit Collectors / Nozzles
- Remaining Work Expectations



### Kogan Creek – Chinchilla - Details

•Kogan Creek Power Station is the newest addition to CS Energy's diverse portfolio ( December 2007). 750 MW, single shaft – super-



750 MW, single shaft – supercritical - Benson Boiler – coal.

**Direct Dry cooled** 

Oxygen treatment – S Steel LP Heaters – C Steel HP heaters

Main steam – 27.1 MPa – 546 °C

Reheat - 566 °C

Econ Inlet Temp 271 °C

2x 50% Consep Polishers cartridge pre-filters – now ammonia cycle

Pre-service HF chemical clean ACC Hot clean (similar to steam blows)

Rifled spiral tubes ASTM A213 T12 (1Cr<sup>1</sup>/<sub>2</sub>Mo)

Condenser vacuum – 20 kpa at 25C

### Kogan Cycle Diagram



# **ACC Construction**

- Steam Ducts, Headers and Pipes Mild Steel
- Condenser and Dephlegmator tubes Mild Steel tubes, external diameter is aluminised 0.5 mm wt, elliptical tube shape, Aluminium cooling fins.
- Single tube row



### Kogan FAC – Iron Transport Issues

- Metal loss within the ACC condensate system
- Particulate loading on filters and polishers
- High iron in feed water prior to chemistry modifications
- Iron Deposition in the boiler / potential for deposition in turbine
- Concern regarding furnace tube creep life and potential for circumferential cracking (minor slagging)
- Need to reduce FAC in the ACC
- Need to consider a boiler chemical cleans as a preventative and management measure – post service clean Sept. 2009.
- Filter and polisher performance critical high condensate temperatures cause polisher trip (70 C) and filter trips (80 C)



### **Implemented Solutions**

- Increased pH to reduce two phase FAC corrosion within the ACC and heater drains system 9.0 to 9.5 then to 9.8
- Operate condensate polishers in ammonia cycle to allow for increased condensate pH.
- Increase polisher thermal trip set point (from 62.5°C to 70 °C).
- Improved condensate filter performance by replacing the filter cartridges



### **Kogan Creek Chemistry History**

Date	рН	Average Iron before prefilters	Average iron after prefilters	Average iron after polishers	Comments		
Commissioning	9.0	72ppb	35ppb	12ppb	During commissioning and early operation		
June 2008	9.5	22ppb	11ppb	4ppb	Polishers changed to ammonia cycle.		
October 2008 (8000hr outage)	9.5	22ppb	<1ppb	<1ppb	Prefilter elements changed to from Pall spiral wound 5µm (nominal) to Graver pleated 5µm (absolute.)		
April 2009	9.8	10ppb	<1ppb	<1ppb	Current operation		



#### Kogan Creek - Condensate Before Pre-filters Iron Monitoring



#### Kogan Creek - Condensate After Polisher Iron Monitoring



#### Kogan Creek - Before Economiser Iron Monitoring



#### August 2011 Outage (3 <sup>1</sup>/<sub>2</sub> Years since commissioning)









# 2009 KOGAN CHEMICAL CLEAN DATA – noting minor Aluminium and Zinc levels

Metal	Final Clean Solution Concentration mg/L	Quantity Removed Kg of Metal		
Iron	8170	817		
Chromium	17	1.7		
Molybdenum	10.2	1.02		
Aluminium	21	2.1		
Copper	<dl< td=""><td>-</td><td></td></dl<>	-		
Tin	<dl< td=""><td>-</td><td></td></dl<>	-		
Zinc	21	2.1		
Suspended Solids	650	65		



### **Turbine Details**

#### Turbine, Types

Four-casing condensing turbine with reheat		
Single-flow HP turbine with 17 reaction stages including 1 low reaction stage and 16 twisted drumstages		
Double-flow IP turbine: with 16 reaction stages per flow including 1 low reaction stage per flow and 15 twisted drumstages per flow		
2 Double-flow LP turbines with 6 reaction stages per flow, including 3 twisted drumstages per flow and 3 standard stages per flow	[···-	



### **Turbo-Generator Overview**







### Kogan Turbine Performance Test Results

Performance Losses in Comparison with the 2010 and 2007 Tests	Equivalent Generation Capacity Loss (MWs)	<i>Likely</i> Recovery from HP Turbine Overhaul (MWs)	<i>Optimistic</i> Recovery from HP Turbine Overhaul (MWs)	
HP Turbine Efficiency Degradation	3.8	1.9	2.5	
HP Turbine Fouling (Likely cause for main steam mass reduction)	14.7	10	14.7	
IP Turbine Efficiency Degradation	4.7	0	0	
HP Turbine Valves	6.3	0	0	
Unaccounted	0.3	0	0	
Total	29.8	11.9	17.2	



### Work Scope for August 2011 Major Overhaul

- HP Turbine
- Video probe inspection of the inlet region
- Video probe inspection of the exhaust region
- Visual inspection of the exhaust region<sup>A</sup>
- Sample the exhaust region deposits Chemical, microscopic and physical analysis
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- IP Turbine
- Video probe inspection of the inlet region
- Video probe inspection of the exhaust region (Generator and turbine sides)
- Visual inspection of the exhaust regions<sup>B</sup>
- Sample the exhaust region deposits Chemical analysis
- •
- LP Turbine
- Visual inspection of the inlet region<sup>C</sup>
- Sample the inlet region deposits Chemical analysis
- Visual inspection of the exhaust region<sup>D</sup>
- Sample the exhaust region deposits Chemical analysis



### HP Exhaust 2009



### HP Exhaust - 2011





### HP Turbine – 2011 – Exhaust 17





### **Kogan IP Turbine**





### Kogan LP Exhaust

LP Turbine Exhaust Region – 2011 inspection





Picture 13

#### **KOGAN TURBINE HP BLADE DEPOSITS - AUG 2011**

Sample	% Fe203	%AI203	%CuO	% Cr203	% C	% CaO	%P2O5	%SO3	Total
Location									
HP Turbine	44.17	18.58	10.37	6.19	16.7	2.14	1.17	0.69	100
Last Blade									
Cold reheat	90.89	2.16	1.18	3.16	2.1	0.5	-	-	100
Pipe Scan 1									
Cold reheat	79.06	14.8	1.24	1.66	2.69	.56			100
Pipe Scan 2									
Cold reheat	82.86	3.97	2.39	3.77	6.6	.39			100
Pipe Scan 3									

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Unfortunately not sufficient HP sample for XRD analysis

\*Carbon – thought to be due to SEM/EDA mounting technique

### Consequences

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- Significant boiler tube iron deposition identified during the plant's 8000hr outage (October 2008)
- Boiler chemical clean conducted to remove deposits (August 2009)
- Deposits of Iron confirmed on last and final 3 stages of the HP turbine (August 2011) – significance? – contribution to MW loss? – significance and source of AI, Cu and Cr.
- Minor Iron deposits found at IP and LP turbine inlets or exhausts (August 2011).
- Further heat rate tests and investigations to be conducted
- Further inspections at mini overhaul 2013
- Planning for first HP cover lift complex costly ?

### **Questions / Investigations**

- What is the source of the Aluminium and copper – is it historical – is it still continuing.
- Detailed water sampling program initiated – method being developed
- Detailed metal mapping of the system undertaken – no answers from study.
- Source of the Aluminium and copper is still not clear



### **Comments on SEM / EDA Results**

 XRD Analysis of Cold reheat 2 sample (14.8 % Al2O3) : showed the presence of fine grained Al2O3 representative of the deposited Al2O3 on the HP turbine blades.





### **Kogan De-aerator Al Deposit**



#### View000

001

#### cmmclient 1/1

Title	: IMG2	
Instrument Volt Mag. Date Pixel	: : 15.00 kV : x 200 : 2011/10/18 : 1280 x 960	-



Mass% Error% Mol% Compound Mass% Cation

87.02 12.98 100.00

14.61 1.39 16.00

91.31 Al203 8.69 Fe203 100.00

Acquisition Parameter Instrument : 6460(LA : 6460(LA) Acc. Voltage : 15.0 kV Probe Current: 1.00000 nA PHA mode : T3 : 49.08 sec Real Time Live Time : 49.00 Sec Live Time : 39.83 sec Dead Time : 17 % Counting Rate: 3520 cps Energy Range : 0 - 20 keV

К

82.3426 17.6574

JED-2300 Analysis Station

(keV)

1.486 6.398

44.87

46.06 9.08

100.00

0.31 1.60

JEOL

### **Current Questions**

- Source of Aluminium? only known Aluminium metal present – external surface of ACC tubes – all other possibilities also being investigated.
- Source of Copper?
- Significance of AI and Cu and turbine blade deposits? – "glue" – also incorporating Fe
- Current investigations apparent MW loss suspected as being due in part to HP blade deposits.
- Other plant Owner's and ACC suppliers experience

### May 2012 Sampling Program

- Method utilises Sentry sampling rig with millipore filter paper and cation exchange paper – follows ASTM Method D6301-08.
- Initial application of method further testing required to refine the method
- Results very low <<1ppb suggests Al present as an insoluble form at 9.8 pH. Question of the capacity of the cation paper to collect soluble Al at pH 9.8



### May 2012 – Sampling Program – Cation Paper

- Filter papers supplied as filters collected from an iron sampling rig. Filters marked as insoluble were 0.1micron mixed cellulose filter papers. Filter papers identified as soluble were cation exchange chromatography paper.
- Method Ref: ICP/OES in accordance with APHA protocol 3030 K



### May 2012 – Sampling Results

CS Energy Kogan PS Identification	Al	Cu	Fe
Metal concentrations in original samples			
(in nanograms per litre)	ng/L	ng/L	ng/L
Before Polisher 1-4/5/12 Insoluble	3.3	1.4	877.3
Before Polisher 1-4/5/12 Soluble	2.6	1.4	21.7
BP 8-15/5/12 Insoluble	33.4	32.4	4375.6
BP 8-15/5/12 Soluble	4.0	0.3	10.8
BP 15-22/5/12 Insoluble	17.0	12.5	7485.0
BP 15-22/5/12 Soluble	3.7	0.3	20.6
BE 15-22/5/12 Insoluble V 2599L	1.6	0.8	30.1
BE 15-22/5/12 Soluble V 2599L	1.0	1.3	1.2
BE 22-29/5/12 Insoluble V 2510L	0.3	0.1	7.6
BE 22-29/5/12 Insoluble V 2510L	0.9	0.9	0.8



### July 2012 – Sampling – Resin Column

- Filter papers supplied as filters collected from an iron sampling rig. Filters marked as insoluble were 0.1 micron mixed cellulose filter papers. Samples identified as soluble were passed through 20g of cation exchange resin and eluted with 250mL of 1+9 HCl. Method Ref: ICP/OES in accordance with APHA protocol 3030 K.
- Results suggest Al is present in soluble form

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### July 2012 – Sampling Results

CS Energy Kogan PS Identification	Al	Cu	Fe
Metal concentrations in original samples			
(in nanograms per litre)	ng/L	ng/L	ng/L
Before Polisher 10-11/7/12 Insoluble	10.2	6.1	2685.0
BP 10-11/7/12 Soluble	108.7	9.4	12.0
BP 11-12/7/12 Insoluble	13.9	2.5	2731.0
P 11-12/7/12 Soluble	310.6	9.4	27.4
BP 13-17/7/12 Insoluble	3.8	3.9	32 <mark>74.0</mark>
BP 13-17/7/12 Soluble	30.2	6.8	13.9
BP 19-23/7/12 Insoluble	3.7	3.1	3987.0
BP 19-23/7/12 Soluble	26.7	2.2	5.2
BP 26-30/7/12 Insoluble	4.3	4.0	5260.0
BP 26-30/7/12 Soluble	22.4	6.7	10.4



### **ASME Code Case 2328-1. - 304 SS** super H - SH and RH tubing C: .07-.13 Mn: 1.00 max Ni: 7.50 - 10.50 Cr: 17.00 - 19.00 Cu: 2.50 - 3.50 Nb: .30 - .60 N: .05 - .12 Al .003 - .030 B: .001-.010 energy

### Materials – WB 36

C 0.10-0-17 Si 0.25-0.50 Mn 0.80-1.20 P 0.025 max S 0.020 max Cr 0.30 max Ni 1.00-1.30 Cu 0.50-0.80 Mo 0.25-0.50 Nb 0.015-0.045 N 0.020 max AI 0.015-0050

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 The CFH side feedline and the BFP Min flow system are manufactured from WB36 steel pipe which contains Cu

### **Deposit Collector – Sampling Nozzle**

Collector

Nozzle





### Deposit Collector – Sampling Nozzle

- Turbine inspection opportunities are limited and blade cleaning opportunities are very irregular.
- Hence;
  - Separate to unit cycle monitoring for Aluminium and copper levels to determining whether deposition is still occurring
  - CS Energy are also investigation the use of deposit collectors and / or converging sampling nozzles as a possible real time method to measure deposition.
- Operating experience is sought from other plant owners



### Remaining Challenges – Expectations - 1

- Other experiences with aluminium blade deposits
   SEM / XRD analysis turbine inspections
- Support for more scientific data on the thermodynamics of Aluminium in the unit cycle – form, transport and deposition mechanisms
- Comments on sampling and analysis of FW steam for aluminium
- Experience with turbine deposit collectors
- Comments on the effectiveness of condensate filters and polishers (ammonia and H cycle) on
   Impigum removal

### Remaining Challenges – Expectations - 2

- Identification of shared interest in further R& D – IAPWS, EPRI and ACCUG.
- Identification of an aluminium limit in FW and steam for sub-critical and super-critical units
- Comments on turbine blade cleaning methods

   in-situ cover on cover off
- Comments on aluminium in boiler tube deposits - is it significant and what is the potential impact of chemical cleaning – does it require special cleaning consideration.





### SHARED EXPERIENCE AND COMMENTS

THANK YOU